NEW HETEROCYCLIC SYSTEM — BENZIMIDAZO[1,2-c] PYRIDAZINO[4,3-e]-1,2,3-TRIAZINE

Yu. M. Volovenko

Diazotization of 6-amino-5-(6-benzimidazol-2-yl)-3-methyl-1-(4-tolyl)-4-pyridazinone (I) [1] using sodium nitrite in acetic acid with subsequent heating of the reaction mixture for 1 h is accompanied by N-azo-coupling involving the resultant diazonium group and favorably located nitrogen atom of the benzimidazole system, leading to closure of a triazine ring. This reaction yields a new heterocyclic system, namely, benzimidazo[1,2-c]pyridazino[4,3-e]-1,2,3-triazine (II).

10-Methyl-8-(4-tolyl)-8,11-dihydrobenzimidazo[1,2-c]pyridazino[4,3-e]-1,2,3-triazin-11-one (II), mp 239-240°C (from DMF), R_f 0.62 (Silufol UV-254, 9:1 chloroform—methanol 9:1). IR spectrum (KBr): 3140-3080 (=C-H), 2910 (-CH₃), 1635 (C=O), 1610, 1570, 1510 cm⁻¹ (N=N, C=C, C=N). PMR spectrum in DMSO-d₆ at 100 MHz: 2.37 (3H, s, CH₃-C₆H₄), 2.46 (3H, s, CH₃), 7.4-7.8 (6H, m, arom), 8.11 (1H, d.d, $J_1^2 = 7$, $J_1^3 = 2$ Hz, 1-H), 8.28 ppm (1H, d.d, $J_4^3 = 7$, $J_4^2 = 2$ Hz, 4-H). The yield of II was 68%. Found: N, 24.7%. Calculated for C₁₉H₁₄N₆O: N, 24.6%.



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Taras Shevchenko Kiev University, 252601 Kiev, Ukraine. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 6, pp. 853-854, June, 1997. Original article submitted December 20, 1996.